GPAT-2021

PY - PHARMACEUTICAL SCIENCES

- 1. Hemophilia A is caused due to the reduction in the quantity or activity of which of the following clotting factors?
- (a) Vitamin K
- (b) Clotting factor IV
- (c) Clotting factor VIII
- (d) Clotting factor IX
- 2. Dynamic Vapour Sorption (DVS) isotherm is a plot of A versus B
- (a) A Crystallization Peak Energy (J/g) B Percentage Relative Humidity
- (b) A Percentage Change in Mass B Percentage Relative Humidity
- (c) A Crystallization Peak Energy (J/g) B Percentage Change in Mass
- (d) A Crystallization Peak Energy (J/g) B Amorphous Content
- 3. Which one of the following is NOT an endogenous opioid peptide?
- (a) met enkephalin
- (b) leuenkephalin
- (c) dynorphin
- (d) beta lipotropin

4. Match the following vaccines under Column I with their respective content under Column II.

Column I	Column II
Vaccine	Content
(P) OPV	(I) Killed bacteria
(Q) Typhoid (TAB)	(II) Live bacteria
(R) Tetanus	(III) Killed virus
(S) BCG	(IV) Live virus
	(V) Toxoid

Choose the correct answer from the options given below:

- (a) (P)-(III), (Q)-(V), (R)-(I), (S)-(II)
- (b) (P)-(1), (Q)-(II), (R)-(V)
- (c) (P)-(IV), (Q)-(I), (R)-(V), (S)-(II)
- (d) (P)-(II), (Q)-(III), (R)-(IV), (S)-(1)
- 5. The size of aqueous filled pores is °A
- (a) 1-3
- (b) 11-18
- (c) 19-25
- (d) 4-10
- 6. Given below are two statements, one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A): The Ayurvedic drug Asava contains self-generated alcohol.

Reason (R): For preparation of Asava the extract of drugs is subjected to a process of calcination. In light of the above statements.

Choose the most appropriate answer from the options given below:

- (a) Both (A) and (R) are correct and (R) is the correct explanation of (A)
- (b) Both (A) and (R) are correct but (R) is NOT the correct explanation of (A)
- (c) (A) is correct but (R) is not correct
- (d) (A) is not correct but (R.) is correct
- 7. The following classification of herbal drugs gives better understanding of biosynthetic pathway in the drug
- (a) Botanical
- (b) Chemotaxonomical
- (c) Taxonomical
- (d) Chemical
- 8. The split up of time cycle of cardiac cycle
- (a) VS-0.3 sec VD-0.5 sec AS-0.1 sec AD-0.7 sec
- (b) VS-0.5 sec VD-0.3 sec AS-0.7 sec AD-0.1 sec
- (c) VS-0.3 sec VD-0.5 sec AS-0.5 sec AD-0.3 sec
- (d) VS-0.5 sec AD-0.3 sec AS-0.3 sec AD-0.5 sec
- 9. Conversion of pseudostratified columnar epithelium in chronic bronchitis and bronchiectasis to columnar type epithelium is an example of which type of metaplasia?
- (a) Squamous metaplasia
- (b) Columnar metaplasia
- (c) Osseous metaplasia
- (d) Cartilaginous metaplasia

- 10. The majority of drugs that least binded to antivascular tissue is(a) Liver(b) Muscle(c) Lungs
- (d) Kidneys
- 11. Distribution is perfusion rate limited in case of...... drugs
- (a) Ionic
- (b) Lipophilic
- (c) Polar
- (d) Water soluble

12. Match the types of terpenes under column I with their respective examples under Column II

Column I Types of terpenes	Column II Examples
(P) Monoterpenes	(I) Lycopene
(Q) Diterpenes	(II) Camphor
(R) Sesquiterpenes	(III) Taxol
(S) Tetraterpenes	(IV) Zingiberene

- (a) (P)-(II), (Q)-(I), (R)-(IV), (S)-(III)
- (b) (P)-(II), (Q)-(III), (R)-(IV), (S)-(I)
- (c) (P)-(IV), (Q)-(I), (R)-(II), (S)-(III)
- (d) (P)-(I), (Q)-(II), (R)-(IV), (S)-(III)

13. Match List I with List II:

List I	List II
Ingredients Found in Rubber Closures	Example
(P) Activator	(I) Zinc dibutyldithiocarbamate
(Q) Curing agent	(II) Stearic acid
(R) Accelerator	(III) Carbon black
(S) Fillers	(IV) Neoprene
	(V) Peroxides

- (a) (P)-(II), (Q)-(V), (R)-(I), (S)-(III)
- (b) (P)-(V), (Q)-(IV), (R)-(II), (S)-(I)
- (c) (P)-(IV), (Q)-(II), (R)-(III), (S)-(V)
- (d) (P)-(III), (Q)-(IV), (R)-(II), (S)-(I)
- 14. Long-term use of thiazide diuretics in hypertension can produce all of the following EXCEPT?
- (a) Hyperkalemia
- (b) Hyperuricemia
- (c) Hyperlipidemia
- (d) Hyperglycemia
- 15. The USP defines As liquid preparations of vegetable drugs, containing alcohol as a solvent or as a preservative, or both, so made that, unless otherwise specified in an individual monograph, each milliliter contains the therapeutic constituents of 1 g of the standard drug that it represents.

- (a) Tinctures
 (b) Extracts
 (c) Infusion
 (d) Fluid extracts
 16. The part of prescription that means 'take thou is
 (a) Subscription
 (b) Inscription
 (c) Superscription
 (d) Signatura
 - 17. Amine group protection during peptide synthesis can be accomplished by converting the amine to
 - (a) 9 Fluorenylmethylcarbamate
 - (b) 1,3-Dithiane
 - (c) Tetrahydropyranyl ether
 - (d) Dimethylacetal
 - 18. The $\beta 1$ and $\beta 2$ adrenergic receptor subtype agonist acting at atrioventricular node produces following responses
 - (a) Increases contractility and conduction velocity
 - (b) Increases automaticity and conduction velocity
 - (c) Increases contractility and automaticity
 - (d) Increases conduction velocity and heart rate

19. Hospital formulary consists of
(a) List of hospital staff
(b) List of drugs present in the hospital
(c) List of equipments
(d) List of patients
20. The ring structure present in strychnine is:
(a) Indole
(b) Purine
(c) Imidazole
(d) Phenanthrene
21. In case of solvates the following are examples of Class I residual solvents that have been classified by the ICH.
solvents that have been classified by the ICH.
solvents that have been classified by the ICH. (a) Benzene, carbon tetrachloride and 1, 2-dichloroethane
solvents that have been classified by the ICH. (a) Benzene, carbon tetrachloride and 1, 2-dichloroethane (b) Acetonitrile, cyclohexane and toluene
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solvents that have been classified by the ICH. (a) Benzene, carbon tetrachloride and 1, 2-dichloroethane (b) Acetonitrile, cyclohexane and toluene (c) Methanol and N.N-dimethylacetamide (d) Acetic acid, acetone and ethanol 22. Total number of facial bones in the human adult (a) 1
solvents that have been classified by the ICH. (a) Benzene, carbon tetrachloride and 1, 2-dichloroethane (b) Acetonitrile, cyclohexane and toluene (c) Methanol and N.N-dimethylacetamide (d) Acetic acid, acetone and ethanol 22. Total number of facial bones in the human adult (a) 1 (b) 7

- 23. An epoxy ring can be prepared by
- (a) DC C oxidation of alcohol
- (b) Prevost reaction
- (c) Darzens glycidic ester synthesis
- (d) Swern oxidation

24. Match List I with List II:

List-I Equipment	List-II Model / Trade Name / Type
(P) Low shear mixer	(I) Manestry Drycota
(Q) Rotary tablet machine	(II) Chilsonator (Fitzpatrick)
(R) Capsule filling machine	(III) Slated Double Cone Mixer
(S) Compaction mill	(IV) Prosolv (JRS Pharma)
	(V) PCF 1200 Model (Pharma Land)

- (a) (P)-(I), (Q)-(II), (R)-(IV), (S)-(III)
- (b) (P)-(II), (Q)-(III), (R)-(I), (S)-(IV)
- (c) (P)-(III), (Q)-(I), (R)-(V), (S)-(II)
- (d) (A)-(I), (B)-(II), (C)-(IV), (D)-(III)
- 25. The immunity acquired by inoculation of living organism of attenuated virulence is:
- (a) Natural passive immunity
- (b) Passive immunity
- (c) Artificial acquired immunity
- (d) Natural active immunity

- 26. Compendium of pharmaceuticals and specialties is an International drug compendia from
- (a) German Pharmaceutical Association
- (b) Australian Pharmaceutical Association
- (c) Indian Pharmaceutical Association
- (d) Canadian Pharmaceutical Association
- 27. Limit for the following heavy metal/s is/are mentioned in the WHO guidelines for herbal drug evaluation.
- (a) Cadmium
- (c) Mercury
- (b) Lead
- (d) Arsenic

- (a) A. B and D only
- (b) Only D
- (c) Only B and D
- (d) Only A and C
- 28. Which of the following ICH Harmonized Tripartite Guidelines related to stability, provides approaches that allows for bracketing and matrixing designs for stability testing. This is especially important when a new drug product is to be commercialized with multiple strengths in multiple packaging configurations. https://www.pyqonline.com
- (a) Q1F
- (b) Q1D

- (c) Q1B
- (d) Q1A

29. Match the following:

(P) Apoptosis	(I) Progressive degradative action of Enzymes
(Q) Necrosis	(II) Tightly regulated intracellular program in which destined
(R) Anaplasia	(III) Abnormal mass of tissue cells to day
(S) Neoplasm	(IV) Lack of differentiation

- (a) (P)-(II), (Q)-(I), (R)-(IV), (S)-(III)
- (b) (P)-(IV), (Q)-(III) (R)-(II), (S)-(I)
- (c) (Q)-(I), (R)-(II), (S)-(III), (P)-(IV)
- (d) (R)-(II), (S)-(III), (P)-(IV), (Q)-(I)
- 30. To determine the effect of breath holding on the deposition of particles, the sedimentation efficiency, S, is defined as.
- (a) S = (Distance the particle falls during breath holding) ÷ (Mean regional airway diameter)
- (b) S = (Distance the particle falls during breath holding) x (Half the mean regional airway diameter)
- (c) S = (Distance the particle falls during breath holding) + (Twice mean regional airway diameter)
- (d) S = (Distance the particle falls during breath holding') (Mean regional airway diameter)

- 31. Which one of the following is a competitive antagonist at benzodiazepine site of GABA-receptor gated chloride channel
- (a) Muscimol
- (b) Flumazenil
- (c) Picrotoxin
- (d) Beta-carboline (DMCM)
- 32. The initial concentration of a drug decomposing according to first order kinetics is 28 units/ mL. The specific decomposition rate, k, obtained from an Arrhenius plot is 2.09 x 10⁻⁵ hr⁻¹ at room temperature. 25°C. Previous experimentation has shown that when the concentration of the drug falls below 15 units/mL it is not sufficiently potent for use and should be removed from the market. What expiration date should be assigned to this product https://www.pyqonline.com
- (a) Approximately 31,012.15 hours
- (b) Approximately 32,225.56 hours
- (c) Approximately 27,216.12 hours
- (d) Approximately 29,869.23 hours
- 33. HEPA filters are defined as A or more efficient in removing, from the air. B particles generated by vaporization of the hydrocarbon Emory 3004.
- (a) A-99.99% B-0.1 μ .m
- (b) A-99.00% B-0.3 μm
- (c) A-99.99% B-0.3 μ .m
- (d) A-99.00 B-0.1 μ .m

- 34. Some individuals are obsessed with using purgatives regularly as a reflection of their psychological problem. Dangers of purgative overuse include all of the following EXCEPT:
- (a) Flaring up of intestinal pathology like rupture of inflamed appendix
- (b) Fluid and electrolyte imbalance, especially hypokalemia
- (c) Steatorrhoea, malabsorption syndrome
- (d) Protein build up in the body as a result of enteropathy
- 35. Which of the following microbial form listed below, exhibits the highest level of resistance to physical and chemical methods of growth control?
- (a) Naked Virus
- (b) Protozoan cyst
- (c) Vegetative bacterial cell
- (d) Bacterial endospore
- 36. Given below are two statements, one is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A): Gram negative bacteria do not retain the primary stain when washed with alcohol and subsequently stained again with secondary stain.

Reason (R): The outer membrane of gram negative bacteria contains lipopolysaccharides. In light of the above statements.

- (a) Both (A) and (R) are true and (R) is the correct explanation of (A)
- (b) Both (A) and (II) are true but (R) is NOT the correct explanation of (A)

- (c) (A) is true but (R) is not false
- (d) (A) is false but (R) is true
- 37. Match the following under chemical compounds under Column I with their respective chemical tests under Column II

Column I Compounds	Column II Chemical tests
P. Quinine	i. Borntrager's
Q. Digitoxin	ii. Thalleoquin
R. Atropine	iii. Vitali Morin
S. Sennosides	iv. Keller killiani

Choose the correct answer from the options given below:

- (a) P-ii, Q-iv, R-iii, S-i
- (b) P-iii, Q-iv, R-ii, S-i
- (c) P-i, Q-ii, R-iii, S-iv
- (d) P-iv, Q-iii, R-ii, S-i
- 38. Marker compound analysis is one of the useful methods for evaluation of herbal drugs.

The following statement is NOT true about marker compound analysis:

- (a) Used to verify potency or identity of herbal drug
- (b) Marker compounds are always therapeutically active
- (c) Quantification of marker compounds can be carried out by HPLC/HPTLC/GC
- (d) The marker compounds are generally secondary metabolites.

- 39. In Gambir-fluorescin test the petroleum spirit layer shows
- (a) Green fluorescence
- (b) Blue fluorescence
- (c) Yellow fluorescence
- (d) Red fluorescence
- 40. Which one of the following plexus is a part of the central nervous system in the brain and consists of capillaries, ventricles, and ependymal cells?
- (a) Celiac plexus
- (b) Choroid plexus
- (c) Cervical plexus
- (d) Brachial plexus

41. Match List I with List II:

List I Name of Meterial	List II Remark
(P) Brasses	(I) Copper-Tin Alloy
(Q) 18/8 Stainless steel	(II) Rigid material
(R) Keebush	(III) Chromium and Nickel
(S) Bronzes	(IV) Flexible material
	(V) Copper-Zinc alloys

- (a) (P)-(V), (Q)-(III), (R)-(II), (S)-(I)
- (b) (P)-(III), (Q)-(II), (R)-(1), (S)-(V)
- (c) (P)-(IV), (Q)-(II), (R)-(I), (S)-(III)
- (d) (P)-(II), (Q)-(IV), (R)-(V), (S)-(III)

42. For depot injections, in which drug is slowly released over weeks or months, Which bioequivalence study design is used?
(a) Multiple dose study design
(b) Replicated crossover study design
(c) Nonreplicate, parallel study design
(d) Randomized crossover design
43. Which one of the following is unorganized drug?
(a) Rauwolfia Root
(b) Acacia Gum
(c) Senna Leaves
(d) Cinchona Bark
44. For solid oral drug products, a change in the concentration of which of the following excipients is more likely to influence the bioavailability of a drug.
(a) Starch (b) Magnasium stearate
(b) Magnesium stearate(c) Microcrystalline cellulose
(d) Lactose
45. Which among the following nuclei can exhibit NMR spectrum
(a) ²⁸ Si
(b) ¹⁶ O
(c) ¹⁴ C
(d) ¹⁴ N

46. Match List I with List II:

List-I Term or Phrase	List-II Meaning
(P) Gutturi	(I) Into the left ear
(Q) In oculum sinistram	(II) Into the right eye
(R) In aurem sinistram	(III) For the left eye
(S) In oculum dextrum	(IV) To the throat

- (a) (P)-(IV), (Q)-(III), (R)-(I), (S)-(II)
- (b) (P)-(II), (Q)-(I), (R)-(IV), (S)-(III)
- (c) (P)-(III), (Q)-(IV), (R)-(I), (S)-(II)
- (d) (P)-(I), (Q)-(III), (R)-(II), (S)-(IV)
- 47. Application for the issue of a permit to import small quantities of drugs for personal use are shall be made in
- (a) Form 12
- (b) Form 12A
- (c) Form 12AA
- (d) Form 13
- 48. How many molecules of acyl-CoA are produced in beta-oxidation of palmitic acid
- (a) Six
- (b) Seven
- (c) Eight
- (d) Nine

- 49. The properties of solutions containing surface active agents change sharply over a narrow concentration range and are called as
- (a) Critical micelle concentration
- (b) Ionic concentration
- (c) Hydrogen ion concentration
- (d) Surface tension
- 50. One of the following apparatus is used to determine particle size by the gravity sedimentation method
- (a) Pyknometer
- (b) Ostwalds viscometer
- (c) Anderson apparatus
- (d) Friabilator
- 51. Class of enzymes which use oxygen as a hydrogen acceptor are known as
- (a) Dehydrogenases
- (b) Oxidases
- (c) Cytochromes
- (d) Nicotinamide coenzymes
- 52. The artificial or synthetic seeds are prepared by the following process
- (a) Somatic hybridization and culture
- (b) Somatic embryogenesis and immobilization
- (c) Microprojectile and transformation
- (d) Artificial pollination and embryogenesis

53. Which of the following statement is the correct description of product positioning?

- (a) Selecting one or more segment to enter
- (b) Occupying distinctive position in the mind of customer
- (c) Identifying group of buyers with common need
- (d) Distinguishes company's product form competitor's product

54. To monitor the safety of the new drug under actual conditions of use in large number of patients has been classified as Clinical trials.

- (a) Phase III
- (b) Phase II
- (c) Phase IV
- (d) Phase I

55. In Mier's theory of supersaturation, solubility chart displays the profile of curve with positive and very low slope in

- (a) KNO₃
- (b) NaCl
- (c) MnSO₄
- (d) None of the above

56. Which of the following is a half-reaction?

(a)
$$Zn + Cu^{2+} \rightarrow Zn^{2+} + Cu$$

(b)
$$H^+ + OH^- \rightarrow H_2O$$

(c)
$$Ag^+ + e^- \rightarrow Ag$$

(d)
$$Ag^+ Cl^- \rightarrow AgC1$$

- 57. Which one of the following calcium channel blockers used as antianginal agents has the highest possibility to produce a negative inotropic effect. AV nodal block. Bradycardia, decreased cardiac output, constipation, edema and is contra-indicated in sick sinus syndrome?
- (a) Nifedipine
- (b) Amlodipine
- (c) Diltiazem
- (d) Verapamil
- 58. Given below are two statements, one is labelled as Assertion (A) and the other is labelled as Reason (R)

Assertion (A): Quillaia tincture is included in Potassium Citrate Mixture BPC.

Reason (R): Because Quillaia tincture act as an emulgent in Potassium Citrate Mixture BPC, and prevent separation of lemon oil as a surface layer when the lemon spirit is added. In light of the above statements.

- (a) Both (A) and (R) are true and (R) is the correct explanation of (A)
- (b) Both (A) and (R) are true but (R) is NOT the correct explanation of (A)
- (c) (A) is true but (R) is false
- (d) (A) is false but (It) is true
- 59. Powers of drug inspectors appointed under the Drugs and Cosmetic Act. 1940 are mentioned under which of the following section of the Act?
- (a) 20
- (b) 21

(c) 22
(d) 23
60. p-gp (p-glycoprotein) is highly expressed on the
(a) Lungs
(b) Pancreas
(c) Superficial columnar epithelial cells of ileum and colon
(d) Heart
61. Select the equation that gives the rate of drug dissolution from a tablet
(a) Fick's law
(b) Henderson Hasselbach equation
(c) Noyes Whitney Equation
(d) Michaelis Menton equation
(a) monde menter equation
62. The hydrogen deficiency index for a compound with molecular
formula $C_{13}H_9N20_4BrS$ is
(a) 9
(b) 10
(c) 13
(d) 8
63. Which one of the following drugs that is not absorbed orally, is not a
bronchodilator, does not have any direct effects on smooth muscles yet
is effective in antigen-induced, exercise- induced and irritant-induced
asthma though not all asthmatics respond to it positively?

(a) Salbutamol (b) Montelukast (c) Sodium cromoglycate (d) Terbutaline 64. All the following can determine the protein structure except (a) Mass spectrometry (b) X-ray crystallography (c) High performance liquid chromatography (d) NMR spectroscopy 65. Which of the following is NOT an intensive property? (a) Concentration (b) Density (c) Pressure (d) Volume 66. Desmethyl diazepam is an active metabolite, having an elimination half-life of more than 40 hours, of all the following drugs EXCEPT: (a) Flurazepam (b) Chlordiazepoxide (c) Clorazepate (d) Diazepam 67. The phenomenon of increasing the solubility of weak electrolytes and non- polar molecules by the addition of a water miscible solvent in which the drug has good solubility is called.

(a) Complexation (b) Co-solvency (c) Solubilization (d) Hydrotrophy 68. The system volume is increased when sheared is called (a) Dilatant flow (b) Newtonian flow (c) Plastic flow (d) Pseudoplastic flow 69. The value of R in energy of activation is (a) 1.98 cal/deg.molecule (b) 1.89 cal/deg.molecule (c) 8.91 cal/deg.molecule (d) 9.18 cal/deg.molecule 70. Which one of the following is not a gene insertion technique into a eukaryotic cell (a) Electroporation (b) Iontophoresis (c) Shot gun sequencing (d) Microinjection 71. A condition in which there is severe loss of extra-cellular fluid due to hemorrhage, diarrhea, severe vomiting or sweating followed by shrinkage

of blood volume, reduction of venous inflow to the heart, reduction of cardiac output and fall of blood pressure is known as:

- (a) Hypovolemic shock
- (b) Anaphylactic shock
- (c) Septic shock
- (d) Cardiogenic shock
- 72. Which one of the following alkaloid contains isoquinoline nucleus
- (a) Quinine
- (b) Hyoscyamine
- (c) Piperine
- (d) Papaverine
- 73. Which of the following method is used for the determination of osmotic pressure?
- (a) Beckmann's method
- (b) Berkeley and Hartley's method
- (c) Cottrell's method
- (d) Landsberger's method
- 74. Which is NOT the colligative property?
- (P) Depression of freezing point
- (Q) Elevation of boiling point
- (R) Elevation of vapour pressure
- (S) Osmotic pressure

Choose the correct one of the following
(a) (P), (Q) and (S)
(b) (P) and (Q)
(c) (P) and (S)
(d) (R) only
75. Which of the following is false about HMP shunt?
(a) No ATP formation
(b) Formation of NADPH
(c) Synthesis of ribose - 5 - phosphate
(d) Non-oxidative phase of HMP pathway is irreversible
76. Vigabatrin produces anti-epileptic activity through
(a) Potentiating GABA mediated neuronal inhibition by depressing GABA transporters
(b) Modifying synaptic release of glutamate/GABA by binding to a specific synaptic protein
(c) Inhibiting GABA transaminase enzyme which degrades GABA
(d) Potentiating post synaptic GAGA receptor effect
77. Substitution constant for Meta benzoic acids is denoted by
(a) σ_m
(b) pKa _(H)
(c) ρ _m
(d) pH _m

- 78. Which among the following is not a calibration parameter for UV visible spectrophotometer?
- A. Control of wavelength
- B. Control of absorbance
- C. Resolution
- D. Control of transmitted radiation
- E. Spectral slit width

- (a) A. B and D only
- (b) A, B, C and E only
- (c) A, B and C only
- (d) A and B only
- 79. In healing of excised wound, there is
- (a) Less tissue loss
- (b) Less inflammatory exudate
- (c) Less necrotic tissue to be removed
- (d) more granulation tissue is required
- 80. A hypertensive patient Mr. Zee already receiving a drug 'X' to control his blood pressure was prescribed a tricyclic antidepressant. This resulted in the abolition of the antihypertensive action of 'X'. Which of the following drug could be X.
- (a) Enalapril
- (b) Clonidine
- (c) Atenolol
- (d) Diltiazem

81. A 3-year-old child presented to OPD with the symptoms of influenza Aspirin is contraindicated in the patient because of increased risk of
(a) Gastric bleeding
(b) Thrombocytopenia
(c) Fanconi Syndrome
(d) Reye's syndrome
82. The total number of amino acids in human insulin (total number of amino acids in A and B chain) is (a) 51
(b) 84
(c) 32
(d) 08
83. Mixed micelle of lipids with bile acids and lipid soluble vitamins are absorbed by https://www.pyqonline.com (a) Pancreatic lipase (b) Apo lipoproteins (c) Enterocytes
(d) Chylomicrons
84. A diuretic compound Cyclothiazide has a substituent at C-3 of 1, 2, 4- benzothioadiazine nucleus?
(a) 5-Bornen-2-yl
(b) 5-Norbornen-2-yl
(c) 2-Bornen-5-y1
(d) 2-Norbornen-5-yl

- 85. Which among the following drugs on standing in aqueous solution can dimerize to 2, 5- bis (aminoxy-methyl) 3, 6-diketopiperazine?
- (a) Pyrazinamide
- (b) Ethionamide
- (c) Diethylcarbamazine
- (d) Cycloserine
- 86. Which among the following statement is correct with respect to their mechanism of action?
- (a) Sitagliptin/Vildagliptin: Dipeptidyl peptidase-4 inhibitors
- (b) Dapagliflozine/Canaglifozine: AMPK activators
- (c) Glibenclamide/ Glimepiride: á-Glucosidase inhibitors
- (d) Acarose/Metformin: KATP channel blockers
- 87. Light directed synthesis on silica wafers is a part of
- (a) Bulk synthesis
- (b) Combinatorial synthesis
- (c) Assymetric synthesis
- (d) Inorganic synthesis
- 88. A compound gives two fragments in a mass spectrometer. They are $C_{19}H_7N^+$ and $C_{19}H_5O^+$ to separate peaks due to these fragments the instrument should have a resolution of
- (a) They cannot be separated
- (b) 2.417 x 10-2
- (c) 3.402×10^4
- (d) 1.042×10^4

89.	The wave length at which maximum absorption of radiation takes
pla	ce is known as
(a)	eta_{max}
(b)	$lpha_{\sf max}$
(c)	γ_{max}
(d)	A_{max}
90.	The immunosuppressant action of cyclosporine is due to
(a)	Interference with antigen recognition
(b)	Inhibition of gene transcription of interleukins
(c)	Blockade of tissue response to inflammatory
(d)	Inhibition of cell wall synthesis
91.	Purpose of Seal coating step of Sugar coating is
(a)	Gives smooth surface to the tablet
(b)	Enhances weight of the tablet for coating
(c)	Gives sweet taste to the Tablet
(d)	Prevent moisture penetration into the tablet
92.	Differential manometer has advantage over simple manometer
(a)	Handling is simple
(b)	It can measure smaller pressure difference
(c)	Viscosity can be determined along with pressure difference
(d)	It is used when pressure difference is high

93. Silver nitrate reagent is used for limit test in Indian Pharmacopoeia
for
(a) Lead
(b) Iron
(c) Chloride
(d) Mercury
94. This compound 2, 5, 8, 11, 14, 17, 20, 23, 26-nona-oxaoctacosan-
28-yl p-(butylamino) benzoate is usually known for its activity?
(a) Antifungal
(b) Antitussive
(c) Vitamin
(d) Glucocorticoid
95. Conversion of T4 to T3 inhibition is associated with
(a) Propylthiouracil
(b) Radioactive iodine
(c) Lugol's iodine
(d) Carbimazole
96. Which of the following is non-epithelial malignancy tumor?
(a) Naevus
(b) Adenoma
(c) Lipoma
(d) Fibrosarcoma

- 97. What will be type of proton NMR spectrum of 2, 2,-dimethylpropane(a) Quartet(b) Doublet(c) Triplet(d) Singlet
 - 98. The dehydration of aldol product of two acetaldehyde molecules through E1cb mechanism

affords formation of

- (a) 1-Butenal
- (b) 2-Butenal
- (c) 3-Butenal
- (d) 4-Butenal
- 99. For very weak acidic drugs. With pKa > 8.
- (a) They are unionized at all pH, hence absorption is rapid and independent of GI pH
- (b) They are affected by changes in pH, hence absorption is pH dependent
- (c) They are ionized in the entire pH range of GIT, hence poorly absorbed from GIT
- (d) They are ionized in the gastric pH, hence better absorbed from the stomach
- 100. Which of the following is wrongly matched combination of antiarrhythmic drugs and their class?

- (a) Lidocaine IB
- (b) Disopyramide II
- (c) Diltiazem IV
- (d) Amiodarone III

101. Match the Tablet Diluents with their properties:

List I Tablet Diluent	List II Properties
(P) Lactose	(I) 2 to 4 percent of moisture
(Q) Mannitol	(II) Maillard reaction
(R) Calcium phosphate	(III) Negative heat of solution
(S) Specially dried starch	(IV) Contraindicated for Tetracycline

Choose the correct answer from the options given below:

- (a) (P)-(III), (Q)-(IV), (R)-(I), (S)-(II)
- (b) (P)-(II), (Q)-(IV), (R)-(I), (S)-(III)
- (c) (P)-(III), (Q)-(I), (R)-(IV), (S)-(II)
- (d) (P)-(II), (Q)-(III), (R)-(IV), (S)-(I)

102. Prostaglandin analogous have therapeutic utility in the following EXCEPT

- (a) Impotence
- (b) Peripheral vascular disease
- (c) Palliative treatment of patent ductus arteriosus
- (d) Pulmonary hypertension
- 103. "Wheal and Flare" reaction is characteristic reaction associated with identification of:

- (a) Type-1 hypersensitivity reaction
- (b) Type-2 hypersensitivity reaction
- (c) Type-3 hypersensitivity reaction
- (d) Type-4 hypersensitivity reaction

104. The total number of possible stereoisomers for 3-bromo-2-butanol are

- (a) 2
- (b) 4
- (c) 6
- (d) 8

105. The absorption maxima of ephedrine hydrochloride and chlorocresol are 257 nm and 279 nm respectively and the A (1%, 1 cm) values in 0.1M hydrochloric acid solution are: 9.0 and zero for ephedrine hydrochloride at 257 nm and at 279 nm respectively. For chlorocresol it is 20.0 and 105.0 at 257 nm and at 279 nm respectively. Calculate the concentration of ephedrine hydrochloride in a batch of Ephedrine hydrochloride injection, diluted 1 to 25 with water, giving the absorbance in 1 cm cell of 0.424 at 279 nm and 0.972 at 257 nm.

- (a) 1.010 mg/ml
- (b) 6.25 mg/ml
- (c) 12.5 mg/ml
- (d) 24.75 mg/ml

106. Which of the following is orally active testosterone derivative having progestational activity?

(a) Oxymesterone (b) Dihydrotestosterone (c) Ethisterone (d) Nomegestrol 107. Which of the following drugs can be imported without a license? (a) Drugs specified in schedule H (b) Drugs specified in schedule X (c) Drugs in schedule-C1 required for manufacturing and not for medicinal use (d) Drugs in schedule-C required for manufacturing and for medicinal use 108. Which among the following cations has the lowest molar ionic conductivity in water at infinite dilution and at 25 degree centigrade? (a) Hydrogen ion (b) Lithium ion (c) Sodium ion (d) Silver ion 109. Formation of pyranose structure from glucose leads to formation of diastereoisomers called (a) Anomers (b) Epimers (c) Tautomers (d) Metamers

110. Match the following

(P) Schedule M	(I) Regulations regarding Life period and storage of various drugs
(Q) Schedule FF	(II) Regulations for manufacturing. Premises. Waste disposal, requirements of plant and equipments (Good Manufacturing Practices)
(R) Schedule G	(III) various drugs to be used under the medical supervision
(S) Schedule P	(IV) Standards for Ophthalmic preparations

Choose the correct answer from the options given below:

- (a) (P)-(I), (Q)-(II), (R)-(III), (S)-(IV)
- (b) (P)-(II), (Q)-(IV), (R)-(III), (S)-(I)
- (c) (P)-(IV), (Q)-(II), (R)-(III), (S)-(I)
- (d) (P)-(III), (Q)-(I), (R)-(II), (S)-(IV)

111. For 1, 3-butadiene, there are.....molecular orbitals for electrons

- (a) 2
- (b) 4
- (c) 6
- (d) 8

112. Which of the following drug belongs to class of antibiotic called ansamycins?

- (a) Rifampin
- (b) Streptomycin

(c) Azithromycin (d) Streptozotocin 113. Four-membered fully saturated heterocycle containing one oxygen is called (a) Oxiren (b) Oxiran (c) Oxet (d) Oxetane 114. Preservatives are usually not added to (a) Small volume parenterals (b) Large volume parenterals (c) All parenterals (d) Bacteriostatic water for injection 115. Name the popular system for ocular delivery of pilocarpine. Based on membrance controlled reservoir systems https://www.pygonline.com (a) Ocusert (b) Alzet pump (c) Progestasert (d) Nitrodur 116. Formaldehyde undergoes Grignard reaction to give (a) Primary alcohol (b) Secondary alcohol

(c) Tertiary alcohol (d) Quaternary alcohol 117. Pyridine upon hydrogenation in pre sence of platinum at 25°C and 3 atm pressure yields (a) Piperidine (b) 2, 3-Dihydropyridine (c) Pyrazine (d) 3, 4-Dihydropyridine 118. Glucose upon treatment with nitric acid yields (a) Gluconic acid (b) Glucaric acid (c) Glucitol (d) Heptanoic acid 119 Chitosan possesses which of the following property? (a) Hypocholesterotemic (b) Antimicrobial (c) Wound healing (d) All of the above 120. The effect of activation or deactivation of an incoming functional group to an aromatic compound is strongest at (a) Only ortho position

(b) Only Meta position

- (c) Only para position
- (d) both ortho and para positions
- 121. The relationship between the vapor pressure and the absolute temperature of a liquid is expressed by https://www.pyqonline.com
- (a) Biles Equation
- (b) Lien and Kennon Equation
- (c) Joule-Thomson Equation
- (d) Clausius-Clapeyron Equation
- 122. Which of the following product life cycle' phases, it is indicative of market saturation in the sense that all the potential customers have bought the product except those who have decided not to have it?
- (a) Growth phase
- (b) Introductory phase
- (c) Decline Phase
- (d) Maturity phase
- 123. One of the following functional group makes benzoic acid weaker
- (a) Nitro group
- (b) Hydroxy group
- (c) Chloro group
- (d) Acetyl group
- 124. When one of the reactants is present in such great excess that its concentration may be considered constant or nearly so, the reaction is said to be of

- (a) Second order(b) Pseudo-order(c) Zero order
- (d) Third order
- 125. Which of the following ointment bases is an ingredient of an absorption base?
- (a) Petrolatum
- (b) Lanolin
- (c) Hydrogenated castor oil
- (d) Cetyl alcohol

ANSWER KEY

1-c	2-b	3-d	4-c	5-d	6-c	7-b	8-a	9-b	10-b
11-b	12-b	13-a	14-a	15-d	16-c	17-a	18-b	19-b	20-a
21-a	22-c	23-с	24-c	25-с	26-d	27-a	28-b	29-a	30-a
31-b	32-d	33-c	34-d	35-d	36-a	37-a	38-b	39-a	40-b
41-a	42-c	43-b	44-b	45-d	46-a	47-b	48-c	49-a	50-с
51-b	52-b	53-b	54-c	55-b	56-c	57-d	58-a	59-с	60-с
61-c	62-b	63-c	64-c	65-d	66-a	67-b	68-a	69-a	70-b
71-a	72-d	73-b	74-d	75-d	76-c	77-a	78-b	79-d	80-b
81-d	82-a	83-c	84-b	85-d	86-a	87-b	88-d	89-d	90-b
91-d	92-b	93-с	94-b	95-a	96-d	97-d	98-b	99-a	100-b
101-d	102-с	103-a	104-b	105-d	106-с	107-с	108-b	109-a	110-b
111-b	112-a	113-d	114-b	115-a	116-a	117-a	118-b	119-d	120-d
121-d	122-d	123-b	124-b	125-b					